Synthesis of Some Heterocyclic Compounds Derived from 2-Mercapto Benzoxazole

Firyal W.Askar*,

Huda A. Hassan**

Nahida A.Jinzeel*

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Abstract

New series of 2-mecapto benzoxazole derivatives (1-20) incorporated into fused to different nitrogen and suphur containing heterocyclic were prepared from 2meracpto benzoxazole, when treated with hydrazine hydrate to afford 2-hydrazino benzoxazol (1). Compound (1) converted to a variety of pyridazinone andphthalazinone derivatives (2-4) by reaction with different carboxylic anhydride. Also, reaction of (1) with phenyl isothiocyanate and ethyl chloro acetate afforded 3phenyl-1,3-thiazolidin-2,4-dione-2-(benzoxazole-2-yl-hydrazone) (6). Azomethines (7-10) were prepared through reaction of (1) with aromatic aldehyde, then (7, 8)converted to thaizolidinone derivatives (11, 12). Treatment of (1) with active methylene compounds afforded derivative (13). Reaction of (1) with CS₂ and NaOH gave 1,2,4-triazole derivative (14). Treatment of (1) with p-bromophenancyl bromide afforded another 1,2,4-triazole (15). The reaction of 2-mercapto benzoxazole with chloro acetic acid gave (16) followed by refluxing (16) with ortho-amino aniline giving benzimidazol (17). Moreover, the reaction of 2-mercapto benzoxazole with ethyl chloroacetate afforded (18), and then reaction of (18) with thiosemicarbazide and 4% NaOH leads to ring closure giving 1,2,4-triazole derivative (20). All compounds were confirmed by their melting point, FT-IR, UV-Vis spectra and ¹H-NMR spectra for some of them.

Key words: Benzoxazole, Thiazolidinone and 1,2,4-Triazole.

Introduction:

Benzoxazole nucleus have been reported various types of biological activites antidepressant, such as antifungal, analgesic [1]. antiinflammatory [2], anticancer and antimicrobial [3].Pyridazinone derivatives as important scaffolds in drug discovery with many of their analog have been used in the treatment of various human pathological states Recently it was found that [4]. pyridazinone derivatives could be used as anticonvulsant activity [5].4thiazolidinone derivatives play a vital role due to their wide range of biological activity [6] and industrial applications [7]. It is also well

established that various derivatives of 1,2,4-triazole exhibit broad spectrum of pharmacological properties such as antibacterial and antifungal activites [8].In the current study, we aimed to synthesize new heterocyclic compounds derivative from 2mercapto benzoxazole containing pyridazinone. thiazolidinone, triazole, benzimidazole moieties with predictable biological activites.

Materials and Methods:

Melting points were taken in open capillary tubes on a Gallenkamp melting apparatus and are uncorrected. The IR spectra (KBr Discs) were recorded with a Shimadzu FTIR 8400.

* Department of Chemistry, College of Science, Al-Mustansiriyah University **Department of Chemistry, IbnAlHaytham College of Education, University of Baghdad The UV-Visible spectra were measured in ethanol using a Shimadzu UV-Vis 160-A spectrophotometer. ¹H-NMR spectra were recorded on Bruker model Ultra Shield 300 MHz spectrophotometer, using TMS as the internal standard in DMSO-d6. Starting chemical compounds were obtained from Fluka or BDH companies.

Preparation of 2-hydrazino Benzoxazole (1):

A mixture of 2-mercapto benzoxazole (1.51g, 0.01mole) and hydrazine hydrate (10ml) was refluxed for 3hrs., ethanol (15 ml) was added and refluxed for 4 hrs. The separated precipitate was filtered and washed with cold water and recrystallized from ethanol.

General procedure for preparation of 1-benzoxazole-2-yl-1,2-dihydropyridazine-3,6-dione (2), 2benzoaxazole 2-yl-2,3,4a,5,8-8a hexa hydro phthalazin-1,4-dione(3) and 2benzoaxazole 2,3dihydrophthalazine-1,4-dione (4):

Maleic anhydride or 1,2,3,6tetrahydro phthalic anhydride or phthalic anhydride (0.01 mole) in 30 ml acetic acid was added to hydrazide (1), (1.49g, 0.01 mole) and the reaction was refluxed for 7hrs. then the mixture was poured on crushed ice, the formed solid product was filtered off and recrystallized from pet. ether (40- 60° C).

Preparation of N-pheny-2-benzoaxazole-2- benzoxazole-1-(2H)thione (14):yl-hydrazine carbothioamide (5):To ethanolic sodium hydroxic

A mixture of compound (1) (1.49g, 0.01mole) and phenyl isothiocyanate (1.31 ml, 0.01mole), in absolute ethanol (20ml) was refluxed for 3 hrs and cooled. The solid product was filtered and recrystallized from ethanol.

Preparation of 3-phenyl-1,3thiazolidine-2,4-dione-2-

(benzoaxazole-2-yl-hydrazone) (6):

Ethyl chloro acetate (1.23g, 0.01mole) was added drop wise to a stirred solution of compound (5) (2.84g, 0.01mole) and anhydrous sodium acetate (0.01 mole) in 20 ml ethanol absolute. The reaction mixture was refluxed for 6 hrs. The solid product was filtered and recrystallized from ethanol.

Preparation of Schiff bases (7-10) :

To a stirring solution of compound (1) (0.01 mole) in ethanol absolute (15 ml), the appropriate different aldehyde from (0.01 mole) was added, and then the mixture was refluxed for 6hrs. and cooled to room temperature. The precipitate was filtered and recrystallized from appropriate solvent.

Preparation of thiazolidenones (11, 12):

A mixture of Schiff bases (7, 8) (0.02 mole) and mercapto acetic acid (0.26ml, 0.04mole) in dry benzene (30 ml) was refluxed for 10hrs. the mixture was concentrated and recrystallized from methanol.

Preparation of compound (13):

To a solution of compound (1) (1.49g, 0.01 mole) in ethanol absolute 30 ml ethyl cyano acetate (0.01 mole) was added. The reaction mixture was heated at refluxed for 6hrs. and after cooling the precipitate was filtered and recrystallized from methanol and water.

Preparation of 1,2,4-triazole [4,3-b] benzoxazole-1-(2*H*)thione (14):

To ethanolic sodium hydroxide solution prepared by dissolving sodium (0.01 mole)in ethanol hvdroxide absolute (30 ml), (0.01 mole) of compound (1) and (0.02mole) CS_2 were added. The mixture was refluxed in water bath at 80°C for 10hrs. then allowed to cool down to room poured temperature, into water. neutralized by dilute acetic acid and the solid product was recrystallized from chloroform

Preparation of compound (15) :

A mixture of compound (1) (1.49g, 0.01 mole) and pbromophenylphenacylbromide (0.01 mole) in ethanol absolute (30 ml) was heated under reflux for 5 hrs. and cooled. The solid product was filtered and recrystallized from ethanol.

Preparation of benzoxazole-2-ylmercapto acetic acid (16) :

To (1.51g, 0.01mole) of 2mercapto benzoxazole in (20 ml) of ethanol absolute, (0.01 mole) of KOH was added followed by (0.095g, 0.01mole) of monochloro acetic acid. The reaction mixture was heated under reflux for 8hrs. The hot solution was evaporated under reduced pressure, the solid was filtered, washed with cold distilled water and recrystallized from ethanol.

Preparation of 2-(1*H*-benzimidazol-2yl-thio methyl) benzoxazole (17):

Compound (16) (2.49g, 0.01mole) was refluxed for 10 hrs with o-phenylenediamine (1.08g, 0.01mole) in 4N hydrochloric acid (20 ml). The mixture was neutralized with ammonia to precipitate benzimidazole. The product was filtered and recrystallized from ethanol.

Preparation of ethyl-(benzoxazole-2thio) acetate (18):

Ethyl chloro acetate (1.23g, 0.01mole) was added drop wise to a stirred solution of 2-mercapto benzoxazole (1.51g, 0.01mole) and KOH (0.56g, 0.01mole) in (20 ml) ethanol absolute. The mixture was refluxed for 5 hrs, and the precipitate was filtered, washed with water and recrystallized from chloroform.

Preparation of 2-[(benzoxazole-2-ylthio) acetyl] hydrazine carbothioamide(19):

To stirring solution of compound (18) (2.37g, 0.01mole) in

ethanol absolute (20 ml) was added thiosemicarbazide (0.91g, 0.01mole). the mixture was refluxed for 4hrs. and cooled to room temperature. The cold was filtered and recrystallized from ethanol and water.

Preparation of 5-(benzoxazole-2-ylthio methyl)-(4*H*), 1,2,4-triazole-3thiol (20):

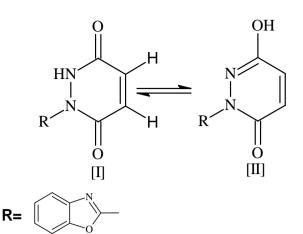
To solution of compound (19) (2.829g, 0.01mole) and (4% of NaOH, 10 ml) was refluxed for 3hrs. the mixture was acidified with dil.HCl and the product was collected and recrystallized from ethanol.

Results And Discussion:

New benzoxazole derivatives containing fused heterocyclic moiety were prepared following the reaction sequence depicted in schemes (1 & 2). 2-mercapto benzoxazole was reacted successfully with hydrazine hydrate. The starting material for the synthesis of targeted compounds is 2-hydrazino benzoxazole (1). Treatment of (1) with carboxylic anhydride, e.g., maleic anhydride, 1,2,3,6-tetraphthalic anhydride and phthalic anhydride gave compounds (2-4).

The structures of compounds (2-4) were confirmed by physical properties and spectral data, which are listed at Table (1). FTIR spectra of (2) show two broad bands at 3455 cm⁻¹ and 3200 cm⁻¹, which assignable to O-H and N-H stretching vibration. The spectrum also shows absorption of C=O of pyridazine ring at 1720 while the C=O stretching of amide occure at 1635 cm⁻¹.

From the above mentioned we can suggest that compound (2) and (3, 4) can be exist in two tautomeric forms: keto [I] and enol forms [II].



The UV Spectrum of compound (3) was obtained in ethanol exhibited the the characteristic bands at (358 nm) responsible for $(n \rightarrow \pi^*)$ transition and (261, 214 nm) due to $(\pi \rightarrow \pi^*)$ transition.

While ¹H-NMR spectra of compound (4) show (δ ppm, DMSO-d₆); 7.3-8.2 (m, 8*H*, Ar-H) and 9.0 (s, 1*H*, NH Phthalazine).

Treatment of (1) with phenylisothiocyanate afforded the corresponding thiosemicarbazide (5). The FT-IT spectra of (5) display C=S stretching band at 1271 cm^{-1} and NH stretching band at 3209 cm^{-1} .

Refluxing of compound (5) with ethyl chloro acetate afforded 4thioazolidone derivative (6). The structure of (6) was confirmed by the precence of C=O stretching band at 1697 cm⁻¹ and C=N stretching at 1643 cm⁻¹ combined with the disappearance of NH₂ stretching band.

Condensation hydridazide (1) with aryl aldehydes in absolute ethanol gave Schiff bases (7-10). The formation of these Schiff bases was indicated by the presence in their IR spectra of the azomethine CH=N stretching at 1600-1644cm⁻¹.

While ¹H-NMR spectra of compound (7) show (δ ppm) 4.2 (s, 1*H*, NH); 7.0-7.8 (m, 8*H*, Ar-H) and 80 (s, 1*H*, =CH).

Moreover, treatment of Schiff bases (7, 8) with mercaptoactic acid in dry benzene gave thiazolidenone derivatives (11, 12), structures of these compounds were confirmed by the presence of C=O stretching band at 1690 cm⁻¹ due to thiazolidinone ring was the characteristic evidence for success of cyclization step.¹H-NMR of compound (12) shows (δ ppm) 4.1 (s, 1*H*, NH); 7.1-7.8 (m, 7*H*, Ar-H), 5.8 (s, 1*H*,N-CH) and 9.5 (s, 1*H*, OH).

Treatment of (1) with active methylene compound such as ethyl cyano acetate produced prazolon derivative (13). In IR spectrum of (13), the carbonyl stretching was observed at 1718 cm^{-1} , whereas the amino group appearance as two bands at 3311 and 3115 cm^{-1} .

¹H-NMR spectra of (13) shows (δppm) 1.2-1.5 (d, 2*H*, CH₂); 5.5 (s, 2*H*, NH₂); (7.3-8.7) (m, 4*H*, ArH).

Condensation of (1) with CS₂ in alkaline medium afforded 1,2,4triazolo [4, 3, a] benzoxazole-3-2*H*thiol (14). The IR spectrum showed N-H stretching absorption in 3267 cm⁻¹ and C=S at 1330cm⁻¹ with a weak absorption near 2660 due to SH stretch because of thiol-thion-tautomeriism, while ¹H-NMR spectra of compound (14) show, δ , 6.3 (s, 1*H*, NH); 7.3-8.7 (m, 4*H*, Ar-H); 13(s, 1*H*, SH).

The interaction of (1) with pbromophenacyl bromide give rise to the formation of 4-(p-bromo phenyl)-6H-1,2,4-triazino [5,4,b] benzoxazole (15). ¹H-NMR spectra of compound (15) show δ 5.5 (s, 1*H*, proton triazine), 6.1-7.7 (m, 8*H*, aromatic NH of trazine).

In order to synthesize benzoxazole-2-yl-mercapto acetic acid (16), the starting material-2-mercapto benzoxazole reacted with monochloro acetic acid. Condensation of (16) with o-phenylenediamine yielded benzimidiazole (17).

Structure of compound (17) was confirmed by spectra data, which showed the disappearance of bands at 3414cm⁻¹ and at 1695 cm⁻¹ attributed to O-H and C=O of carboxylic in compound (16), while ¹H-NMR spectra of compound (17) show δ 4.2

(s, 2*H*, SCH₂); 7.3-7.9 (m, 8*H*, Ar-H); 12.1 (s, 1*H*, NH of amidazole).

Moreover, treatment of 2mercapto benzoxazole with ethyl chloro acetate afforded (18), which displayed C=O stretching band at 1734 cm^{-1} . Refluxing of (18)with thiosemicarbazide in dry benzene give acyl thiosemicarbazide (19), which upon ring closure with 4% NaOH afforded 5-(benzoxazole-2-ylthiomethyl)-4*H*-1,2,4-triazole-3-thiol (20), which exists in a tautomericthiolthione equilibrium as andicated by C=S stretching band at 1236 cm^{-1} and S-H stretch at 2550 cm^{-1} .

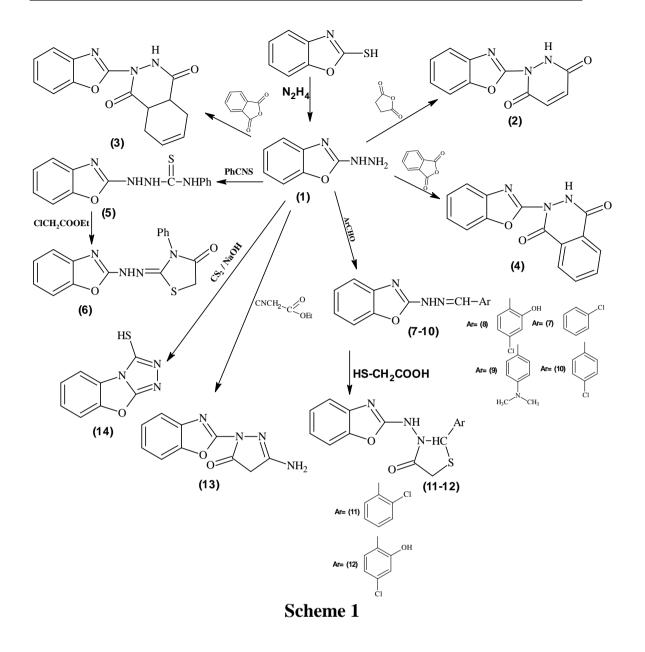
Table(1): Physical Properties and Spectral Data of thePrepared Compounds

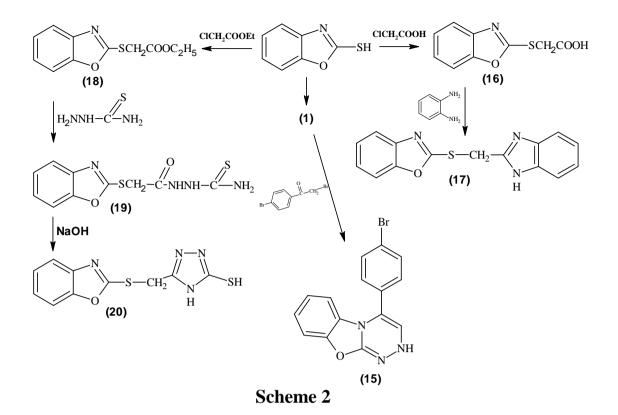
	r repared Compounds						
Com.	Formul	M.p. °C	Yield%	Color	UV	Infrared data (v , cm ⁻¹) KBr	
No.					λ _{max}	disc	
1	C ₇ H ₇ N ₃ O	210-212	74	White	205	3421, 3267 N-H, 3091 C-H ar.	
					343	1639 C=N, C-O-C 1157	
2	$C_{11}H_7N_3O_3$	82-84	55	Pale	202	3450 О-Н, 3221 N-Н, 2930 С-Н	
				brown	250	al., 1710 C=O, 1627 C=O amid	
					389		
3	$C_{15}H_{11}N_3O_3$	110-112	59	Brown	214	3173 N-H, 2916 C-H al., 1718	
					261	C=O, 1655 C=O amid, 3309 O-H	
					358		
4	$C_{15}H_9N_3O_3$	204	57	Nrown	215	3308 О-Н, 3186 N-Н, 3016 С-Н	
		205			356	ar., 1699 C=O, 1626 C=O amid	
5	$C_{14}H_{12}N_4OS$	208	70	Yellow	211	3269, 3171 NH, 3028 C-H ar.,	
		210			302	1271 C=S	
6	$C_{16}H_{12}N_4O_2S$	175	62	Dark	219	3209 N-H, 3024 C-H ar., 2929 C-	
	~ ~ ~ ~ ~ ~ ~ ~	177		yellow	230	H al., 1697 C=O, 1643 C=N	
7	$C_{14}H_{10}N_3OC$	211-213	77	White	218	3184 N-H, 3014 C-H, 1616 C=N,	
	1				263	1049 C-Cl	
0	C U NO	227	70	N 7 11	318	2211 O.H. 2120 N.H. 2020 G.H.	
8	$C_{14}H_{10}N_3O_2$	227	78	Yellow	202	3311 O-H, 3120 N-H, 3020 C-H	
	Cl	228			228 338	ar., 1631 C=N, 1006 C-Cl	
9	C ₁₆ H ₁₃ N ₄ O	182-184	80	Yellow	202	3309 N-H, 3045 C-H ar., 1626	
9	$C_{16}\Pi_{13}\Pi_{4}O$	102-104	80	Tenow	202 350	C=N, 1010 C-Cl	
10	C ₁₄ H ₁₀ N ₃ OC	215-217	82	White	204	3308 N-H, 3026 C-H ar., 1710	
10		215-217	02	white	260	C=0, 1045 C-Cl	
11	C ₁₅ H ₁₂ N ₃ O ₂ S	230-231	70	Yellow	200	3298 N-H, 3095 C-H ar., 2906,	
11	Cl	230 231	,0	1 0110 W	250	2818 C-H al., 1614 C=N	
	01				360	2010 0 11 41, 101 1 0-11	
12	C ₁₆ H ₁₂ N ₃ O ₃ S		72	Yellow	220	3450 О-Н, 3275 N-Н, 1720 С=О,	
	Cl		. –		380	1016 C-Cl	
13	$C_{10}H_8N_4O_2$	220-221	66	White	201	3311, 3115 NH ₂ , 3020 C-H ar.,	
	10 0 . 2				219	2949 C-H al., 1718 C=O, 1631	
					315	C=N	

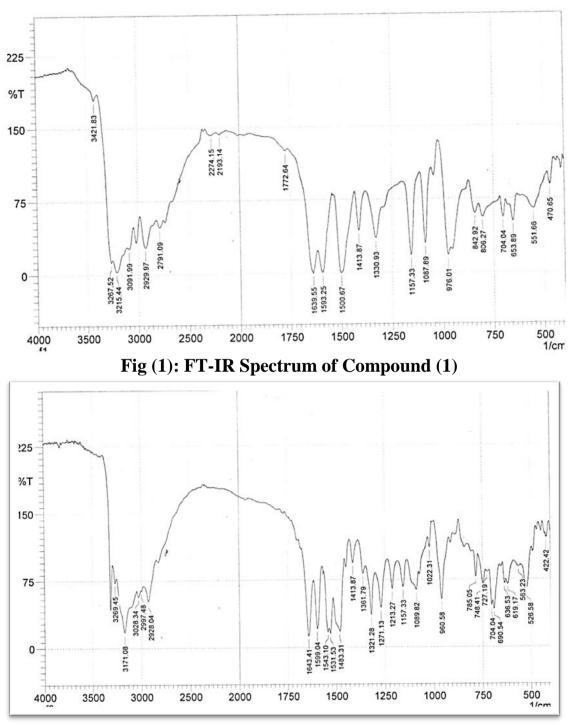
14	C ₈ H ₅ N ₃ OS	100-101	63	Brown	204	3207 N-H, 3084 C-H ar., 1643
14	$C_8\Pi_5\Pi_3OS$	100-101	05	DIOWII	-	
					262	C=N, 1330 C=S
15	$C_{15}H_{10}N_3OB$	74-76	60	Brown	204	3244 N-H, 3016 C-H ar., 1633
	r				387	C=N, 1604, 1570 C=C, 850 C-
						H, 603 C-Br
16	C ₉ H ₇ NO ₃ S	121-122	60	Brown	240	3500-3200 O-H, 3080 C-H ar.,
					330	1695 C=O, 603 C-S
17	$C_{15}H_{11}N_3OS$	250-252	69	Yellow	211	3377, 3306 NH, 3051 C-H ar.,
					250	2960 C-H al., 1620 C=N
					360	
18	$C_{11}H_{11}NO_3S$	211-212	70	White	208	3090 C-H ar., 2935 C-H al., 1734
					360	C=O ester, 1215 C-O, 1639 C=N
19	$C_{10}H_{10}N_4O_2S$	149-151	67	Yellow	203	3327, 3263 NH ₂ , 3158 N-H, 3064
	2				360	C-H ar., 1663 C=O, 1242 C=S
20	$C_{10}H_{10}N_5OS_2$	231-232	55	Brown	215	3170 NH, 2550 SH, 1640 C=N,
					380	1236 C=S

Table(2): Chemical Schiff's ¹H-NMR Spectra

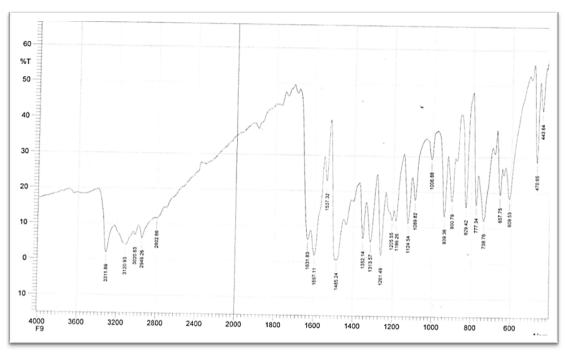
Com.	¹ H-NMR (DMSO-d ₆) δppm
No.	
4	7.3-8.2 (m, 8 <i>H</i> , Ar-H); 9.0 (s, 1 <i>H</i> , NH phthalazine)
7	4.2 (s, 1 <i>H</i> , NH); 7.0-7.8 (m, 8 <i>H</i> , Ar-H); 8.0 (s, 1 <i>H</i> , =CH)
12	4.1 (s, 1 <i>H</i> , NH); 7.1-7.8 (m, 7 <i>H</i> , Ar-H); 5.8 (s, 1 <i>H</i> , N-CH); 9.5 (s,
	1 <i>H</i> , OH)
13	1.2-1.5 (d, 2 <i>H</i> , CH); 5.5 (S, 2 <i>H</i> , NH ₂); 7.3-8.7 (m, 4 <i>H</i> ,Ar-H)
14	6.3 (s, 1 <i>H</i> , NH); 7.3-8.7 (m, 4 <i>H</i> , Ar-H); 13.0 (s, 1 <i>H</i> , SH)
15	5.5 (s, 1H, Proton triazine); 6.1-7.7 (m, 8H, aromatis NH of
	triazine)
17	4.2 (s, 2H, SCH ₂); 7.3-7.9 (m, 8H, Ar-H); 12.1 (s, 1H, NH of
	amidazole)

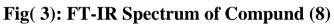


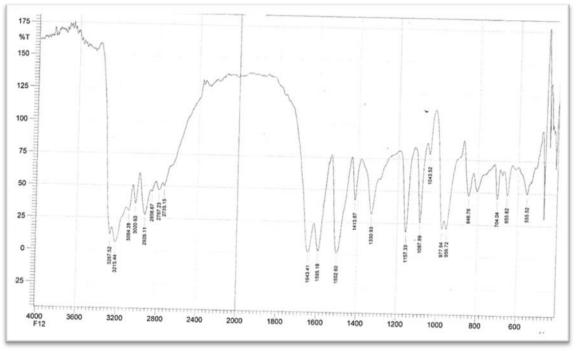




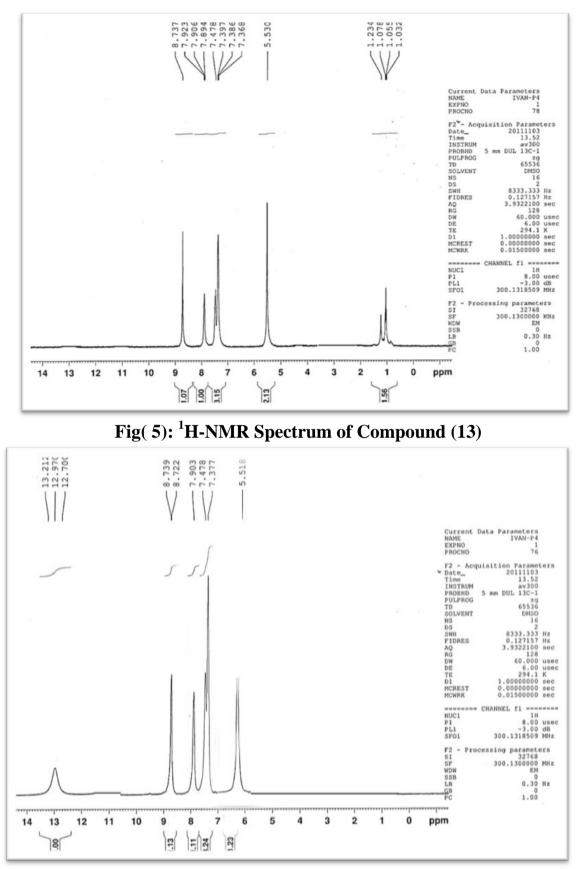


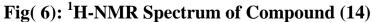






Fig(4): FT-IR Spectrum of Compound (14)





References:

- 1- Lokwani P., Nagori B. P., Butra N., Goyal A., Gupta S. and Singh N., 2011 "Benzoxazole: The Molecule of Diverse Biological Activites", J. Chem. Pharma. Res., 3(3), 302-311.
- 2- Sunila T., Parloop A., 2010 Synthesis and Pharmacological Screening of Some Benzoxazole Derivatives as an Antiflammatory Agents", International Journal of Pharmaceutical Research and Development, 2(24), Issue 9.
- 3- Samia M., A. Fawzia, Soad A., Mona M., Mona H. and Manal A., 2005 "Synthesis of Some Novel Benzoxazole Derivatives as Anticancer Anti-H1V1 and Antimicrobial Agents", European Journal of Medical Chemistry Vol. 40. Issue 9.949-959.
- 4- Vanadma S. A. and Sharma K. V. 2006 Synthesis and Biological Some 3,5-Diaryl-1- BenzothiazolePridazine Derivatives", E-Journal of Chemistry, 6(2), 356-384.
- 5- Chandar K., Asif M., Gary V., Sharma P. and Singh R. 2011 "Synthesis of Different Substituted Pyridazinone Derivatives and Their Anticonvulsant Activity", E-Journal of Chemistry, 8(1), 245-251.
- 6- Rajiv D., Sowan S. K. and Srivastava, S. D. 2010 Synthesis and Antimicrobial Activity of Some 2[(2-Substituted-5-Methyl-1,3-

Thiazalidine-4-One-5-(2-

Methylamino-4-Phenyl-1,3-

Thiazolyl]-1,3,4-Thiazoles", Int. J. Res. Sci., Issue 3, 558-364.

- 7- Srividhya D, Maniunathan S. and Thirumaran S., 2009 "Synthesis and Characterization of New Heterocyclic Liquid Crystals", E-Journal of Chemistry, 6(3), 928-937.
- 8- Sabir H., Jyoti S.and Mohd A. 2008 Synthesis and Antimicrobial Activites of 1,2,4-Triazole and 1,3,4-Thiadiazole Derivatives of 5-Amino-2-Hydroxy Benzoic Acid", E. Journal of Chemistry, 5(4), 963-968.
- 9- Nadia A. 2008Systhesis and Characterization of Novel Azole Heterocycles Based on 2,5-Disubstituted Thiadiazole", Turk J. Chem., 32, 229-235.
- 10- Vinod Kumar P., Mridula U. and Mrinaljni, U. 2005 ."BenzimidazolylQuinolinylMercapto Triazoles as PotentiaAntimicrobal and Antiviral Agents", Acta Pharm., 55, 47-56.
- 11- Williams and Fleming, 1973Spectroscopic Methods in Organic Chemistry", 2nd Ed., McGraw Hill, London.
- 12- Robert Silverstein M., Francis Webster X. and DavedKiemle, J. 2005Spectrometric Introduction of Prganic Compounds", 7th Ed., John Wiley and Sons, USA.

تحضير بعض المركبات الحلقية الغير متجانسة المشتقة من2- مركبتو بنزو اوكسازول

فريال ولى عسكر * هدى احمد حسن * * ناهدة عبد الله جنزيل *

*الجامعة المستنصرية – كلية العلوم – قسم الكيمياء
** جامعة بغداد – كلية التربية ابن الهيثم

الخلاصة

تم تحضير سلسلة جديدة من مركبات حلقية غير متجانسة (1–20) مشتقة من 2-مركبتو بنزو أوكسازول تحتوي على ذرات مختلفة من النتروجين والكبريت وذلك عند معاملة المركب أعلاه مع الهدرازين اللامائي للحصول على 2-هايدرازوبنزو أوكسازول (1). ثم تحويل المركب إلى عدد من مشتقات البيردايازنون وفثالزينون (2–4) بتفاعله مع عدد من أنهدريدات الحوامض الكاربوكسيلية. تفاعل (1) مع فنيل آيزوثايوسيانيت وأنيُل كلورو أستات فتكون (2–4) مع عدد من أنهدريدات الحوامض الكاربوكسيلية. تفاعل (1) مع فنيل آيزوثايوسيانيت وأنيُل كلورو أستات فتكون (2–6) مع عدد من أنهدريدات الحوامض الكاربوكسيلية. تفاعل (1) مع فنيل آيزوثايوسيانيت وأنيُل كلورو أستات فتكون (1) مع عدد من أنهدريدات الحوامض الكاربوكسيلية. تفاعل (1) مع فنيل آيزوثايوسيانيت وأنيُل كلورو أستات فتكون (10). ثم تعاعل (1) مع فنيل آيزوثايوسيانيت وأنيُل كلورو أستات فتكون (10) حضرت من تفاعل (1) مع الأدرياتية، ثم تحول المركبات (7 و8) إلى عدد من مشتقات ثايوزوليدنون فتكون (11 و12). ومعاملة (1) مع مركب المثيلين الفعال أعطت المركب (1). ثم تفاعل (1) مع مركب المثيلين الفعال أعطت المركب (1). ثم تفاعل (1) مع مركب المثيلين الفعال أعطت المركب (13). ثم تفاعل (1) من و22 و 2-5 وريزول (14). ومعاملة (1) مع مركب المثيلين الفعال أعطت المركب (10). ثم تفاعل (1) مع و23 و 25 و (11 و21). ومعاملة (1) مع مركب المثيلين الفعال أعطت المركب (13). ثم تفاعل (1) مع و25 و 25 وريزول (14). ومعاملة (1) مع بارا–بروموفيناسيل برومايد أعطى مد و25 و 25 وريزول (14). ومعاملة (1) مع بارا–بروموفيناسيل برومايد أعطى أيل كاورو أسيتاك أعطى مشتق 1 و2 و 4-ترايزول (14). ومعاملة (1) مع بارا–بروموفيناسيل برومايد أعطى أل يعد مع حامض كلورو أسيتاك أعطى (10)، وعند مع حامض كلورو أسيتاك أعطى (10)، وعند مع حال الركب (10). مع أوريو أوكسازول مع حامض كلورو أسيتاك أعطى (10)، وعند معنيل المركب (13). تم تفاعل (1) مع ثاريميدازول مع حامض كلورو أسيتاك أعطى (10)، وعند معنيل النولي في أيل كلورو أسيتات أعطى (18). ومن ثم تفاعل (11) مع ثايوسيماكاربازايد في محيط قاعدي أدى، وعد أيئيل كلورو أسيتات أعطى مشتق 1، 2، 4-ترايزول (20). وشخصت المركبات المحضرة جميعها عن طريق قياس أيئيل كلورو أسيتات أعطى مشتق 1، 2، 4-ترايزول (20). وشخصت المركبات المحضرة جميعها عن طريق أل ررجة نورجا ويالي راروي أ