

The Effect of Thickness on Some Optical Properties of Sb₂S₃ Thin Films Prepared by Chemical Bath Deposition

*Nadir Fadhil Habubi** *Sami Salman Chiad** *Saba J. Hasan**

Date of acceptance 28 / 2 / 2010

Abstract:

Sb₂S₃ thin films have been prepared by chemical bath deposition on a glass substrate. Absorbance and transmittance spectra were recorded in the wavelength range (30-900) nm.

The effects of thickness on absorption coefficient, reflectance, refractive index, extinction coefficient, real and imaginary parts of dielectric constant were estimated. It was found that the reflectivity, absorption coefficient, extinction coefficient, real part of dielectric constant and refractive index, all these parameters decrease as the thickness increased, while the imaginary part of the dielectric constant increase as the thickness increased.

Keywords: Sb₂S₃, Thin Films, CBD, Optical Properties.

Introduction:

Antimony trisulfide which desorbed as a chalcogenide material has attracted much attention due its unique properties such as high absorption coefficient, optimum band gap ($E_g = 1.8$ eV) [1], high refractive index [2], well-defined quantum size effects [3,4], this material exhibit structural modification when irradiated by light and by an electron beam [5, 6], their photosensitive and thermoelectric properties [7].

Therefore it is worthy to pay more attention to antimony trisulfide as a very promising material for industrial application which could be used in a potential material for applications in photovoltaic structures [8,9], optical data storage devices [10], electronic devices specially in photo conductive targets for the vidicon type of the television camera tubes [11], solar cells [12], rechargeable storage cell [13], resonant laser cavity [14]. Sb₂S₃ has been synthesized by various methods including chemical deposition

method [15], thermal evaporation [16], radio frequency sputtering [17] and chemical bath deposition [18]. From the methods used for the preparation of Sb₂S₃ thin films, the chemical bath deposition method is often preferred because it offers large possibilities to modify the deposition condition so as to obtain films with good structure and physical properties. So the aim of this work is to prepare thin of Sb₂S₃ utilizing chemical bath deposition in order to study the effect of film thickness on some optical parameters of Sb₂S₃.

Material and Methods:

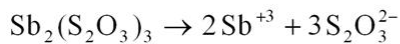
Antimony trisulfide thin films with different thicknesses were deposited on glass substrate using chemical bath deposition technique.

11.5 g of SbCl₃ was dissolved in 50 ml acetone of this solution were placed in 50 ml beaker to which 12 ml of 1M Na₂S₂O₃ and 33 ml of deionised water. The resulting solution was stirred for 3 minutes, the PH of the bath was

* Physics Department, College Of Education ,Mustansiriyah University

measured at 3.5. the glass substrate were using a neutral detergent and then dried well using warm air two substrates were attached to each other and placed vertically in the beaker. This process was repeated for different dip time in order to obtain films with different thicknesses ranging from (1-5) hours, the films thickness were (0.1, 0.3, 0.4) μm .

The chemical reaction for this solution could be written as:
 $\text{SbCl}_3 + \text{Na}_2\text{S}_2\text{O}_3 \rightarrow \text{Sb}_2(\text{S}_2\text{O}_3)_3 + 6\text{NaCl}$



The antimony ions, together with the sulphide ions produced in the bath, condense at the substrate surface forming Sb_2S_3 Thin Film absorbance and transmittance spectra were recorded in the wavelength range (300-900) nm using UV/VIS double beam shimadzu 160A spectrophotometer (Japan).

Results and discussions

Fig. (1) Presents the reflectance (R) in the wavelength range (300-900) nm for different thickness of Sb_2S_3 thin films, with increasing films thickness the reflectance decrease from an average of 0.2 to less than 0.1.



Fig. (1) Reflectance versus wavelength

for the as deposited films. Since the reflectivity was insignificant near the absorption edge, so the following relation could be use for calculation the absorption edge, so the following relation could be use for calculating the absorption coefficient (α) [19]:

$$\alpha = \frac{2.303A}{t} \dots(1)$$

Where (A) is the absorption and (k) the film thickness.

Fig. (2) shows the dependence of α on photon energy, the absorption edge shift toward lower wavelength range (blue shift). The shift in the absorption edge may be attributed to the difference in grain size [20] and /or carrier concentration [21].

The refractive index (n) of the Sb_2S_3 thin films were calculated by the following relation [22]:

$$n = \frac{(1+R)}{(1-R)} + \sqrt{\frac{4R}{(1-R)^2} - K^2} \dots(2)$$

Where (n) is the refractive index and (k) is extinction coefficient and was calculated using the relation [23]:

$$k = \frac{\alpha \lambda}{4 \pi} \dots(3)$$

Where α is the absorption coefficient and (λ) is the incident wavelength.

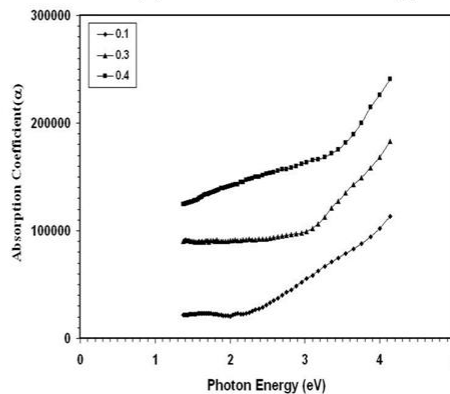


Fig. (2) Absorption coefficient versus photon energy for the as

deposited films.

From Fig. (3) on can easily obtain the extinction coefficient of Sb_2S_3 with different thicknesses it can be seen that the extinction coefficient increases as the thickness increase.

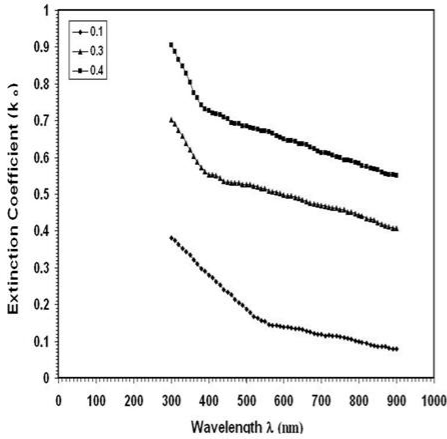


Fig. (3) Extinction coefficient versus wavelength for the as deposited films.

Fig. (4) Shows the dependence of the refractive index of Sb_2S_3 on wavelength, it can be seen that refractive index decreases as the thickness increased. The variation of the optical constants of the films can be correlated with their structure characteristic, the decrease of n and increase of k may be due improvement of stoichiometry [24], the increase in grain size and the decrease is micro strain, this behavior of the refractive index and extinction coefficient with wavelength is in good agreements with studies reported by de Araújo, et al. [25], Rajpure and Bhosale [26], El Zawawi et al. [27]

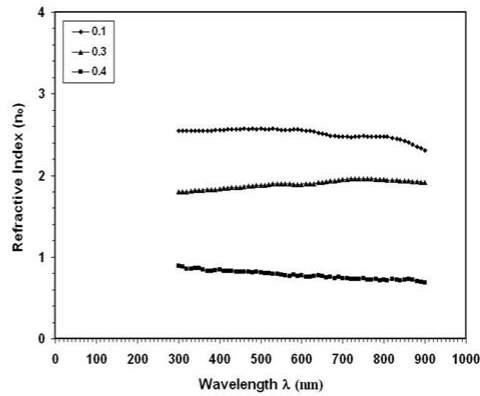


Fig. (4) Refractive indices versus wavelength for the as deposited films.

The dielectric constant (ϵ_T) can be defined as [28]:

$$\epsilon_T = \epsilon_1 - i\epsilon_2 \quad \dots(4)$$

The real (ϵ_1) and imaginary (ϵ_2) parts of the dielectric constant are related to the (n) and (k) values. The (ϵ_1) and (ϵ_2) values can be calculated using the form formulas [29]:

$$\epsilon_1 = n^2 - k^2 \quad \dots(5)$$

$$\epsilon_2 = 2nk \quad \dots(6)$$

Fig. (5) and Fig. (6) Presents the dependence of the real dielectric constant of the Sb_2S_3 thin film on wavelength, it is clearly seen from the fig that both real and imaginary parts decreased as the thickness increase.

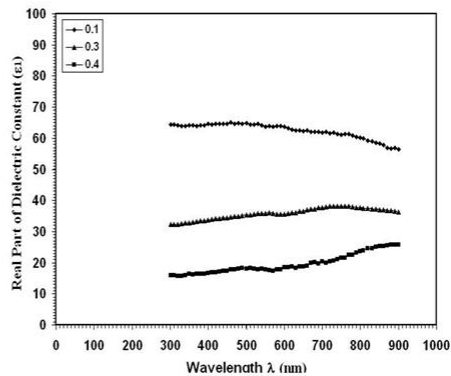
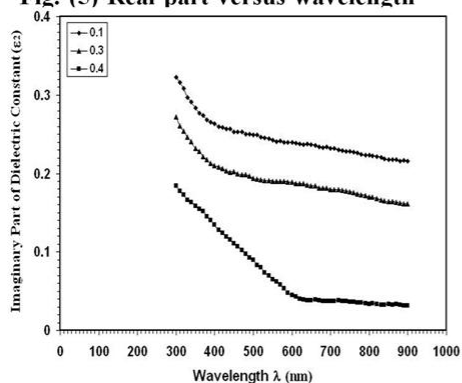


Fig. (5) Real part versus wavelength**Fig. (6) Imaginary part versus wavelength for the as deposited films.****Conclusions:**

- 1- Chemical bath deposition proved to be a featured technique to prepare Sb_2S_3 .
- 2- The value of the absorption coefficient increases slowly at long wavelengths where as it increases rapidly at low wavelength region, this behavior supports the assumption of using these thin films as antireflection coatings in the long wavelength region, while they are used for fabrication light emitting diode in the second region.

References :

- [1] Tigau, N., Gheorghies, C., Rusu, G. I., Condarache Bota, S. 2005. The Influence of The Post Deposition on Some Physical Properties of Sb_2S_3 Thin Films, *Journal of Non-Crystalline Solids*, 351 : 987.
- [2] Krishnan, B., Arato, A., Cardenas, E., Das Ray, T. K. and Castillo, G. A. 2008. on The Structure, Morphology and Optical Properties of Chemical Bath Deposited Sb_2S_3 Thin Films, *Applied Surface Science*, 254 : 3200.
- [3] Salem, A. M. and Soliman Selim, M. 2001. Structure and Optical Properties of Chemically Deposited Sb_2S_3 Thin Films, *Journal of Physics D: Applied Physics*, 34: 12.
- [4] Mane, R. S. and Lokhande, C. D. 2003. Thickness Dependent Properties of Chemically Deposited Sb_2S_3 Thin Films, *Material Chemistry and Physics*, 82 (2): 347.
- [5] Fritzsche, H. 1995. Optical Anisotropies in Chalcogenide Glasses Induced by Band Gap Light, *Phys. Rev. B* 52 : 15854.
- [6] Debnath, R. K. and Fitzgerald, A. G. 2005. Electron Beam Induced Surface Modification of Amorphous Sb_2S_3 Thin Films, *Applied Surface Science*, 243 : 148.
- [7] Roy, B., Chakraborty, B. R., Bhattacharya, R. and Dutta, A. K. 2008. Electrical And Magnetic Properties of Antimony Sulphide Crystals and The Mechanism of Carrier Transport in it, *Solid State Communications*, 25 (11) : 937.
- [8] Savadoge. 1998. Chemically and Electrochemically Deposited Thin Films for Solar Energy Materials, *Solar Energy Materials and Solar Cells*, 52 (3-4) : 361.
- [9] Messina, S., Nair, T. S. and Nair, P. K. 2007. Antimony Sulfide Thin Films in Chemically Deposited Thin Film Photovoltaic Cells, *Thin Solid Films*, 515: 5777.
- [10] Arun, P., Vedeshwar, A. G. and Mehra, N. C. 1997. Laser Induced Crystallization in Sb_2S_3 Films, *Materials Research Bulletin*, 32 (7) : 907.
- [11] Ghosh, G. and Varma, B. P. 1979. Optical Properties of Amorphous and Crystalline Sb_2S_3 Thin Films, *Thin Solid Films*, 60 (1) : 61.
- [12] Messina, S., Nair, M. T. S. and Nair, P. K. 2009. Solar Cells With

- Sb₂S₃ Absorber Films, Thin Solid Films, 517 : 2503.
- [13] Rajpure, K. Y. and Bhosale, C. H. 2000. Sb₂S₃ Semiconductor Spectrum Rechargeable Storage Cell, Material Chemistry and Physics, 64 (1) : 70.
- [14] Arato, A., Cardenas, E., Shaji, S., O'Brien, J. J., Liu, J., Alan Castillo, G., Das, Roy, T. K. and Krishnan, B. 2009. Sb₂S₃:CdS P-n Junction by Laser Irradiation, Thin Solid Films, 517 : 2493.
- [15] Lokhande, C. D., Sakapal, B R., Mane, R. S., Pathon, H. M., Muller, M., Ciersing, M. and Ganesan, V. 2002. XRD, SEM, AFM, HRTEM, EDAX and RBS Studies of Chemically Deposited Sb₂S₃ and Sb₂Se₃ Thin Films, Applied Surface Science, 193 : 1.
- [16] EL-Shazly, A. A., Segam, M. A. M., Samanoudy, M. M., El, Ammar, A. H. and Assim, E. M. 2002. The Effect of Deposition Rate and Heat Treatment on Conduction and Charge Carrier Transport Mechanism in Sb₂S₃ Films, Applied surface Science, 189 : 129.
- [17] Versavel, M. Y., Haber, J. A. 2007. Structural and Optical Properties of Amorphous and Crystalline Antimony Sulfide Thin Film, Thin Solid Films, 515 : 7171.
- [18] Ezema, F. I., Ekwealor, A. B. C., Asogwa, P. U., Ugwuoke, P. E., Chigbo, C., and Suji O, R.U. 2007. Optical Properties and Structural Characterizations of Sb₂S₃ Thin Films Deposited by Chemical Both Deposition Technique, Turk J. Phys., 31 : 205.
- [19] Han, X., Liu, R., Chen, W. and Xu, Z. 2008. Properties of Nanocrystalline Zinc Oxide Thin Films Prepared by Thermal Decomposition of Electrodeposited Zinc Peroxide, Thin Solid Films, 516 : 4025.
- [20] YU, J., Zhuo, X., and Zhao, Q. 2000. Effect of Surface Structure on Photo Catalytic Activity of TiO₂ Thin Films Prepared by Sol - Gel Method, Thin Solid Films, 379 (1-2) : 7.
- [21] Tang, W., and Cameron, D. C. 1984. Aluminum Doped Zinc Oxide Transparent Conductors Deposited by The Sol - Gel Process, Thin solid Films, 238 (1) : 83.
- [22] Llican, S., Caglar, Y., Caglary, M. and YaKuphanoglu, F. 2008. Structural Optical and Electrical Properties of F-Doped ZnO Nanorod Semiconductor Thin Film deposited by Sol - Gel Process, Applied Surface Science, 255 : 2353.
- [23] Xue, S.W., ZU, X. T., Zhou, W. L., Deng, H. X., Xiang X. and Deng, H. 2008. Effects of Post - Thermal Annealing on The Optical Constants of ZnO Thin Film, J. of Alloys and Compounds, 448 : 21.
- [24] Ashour, A., EL-Kadary, N. and Mahmoud, S. A. 1995. Effect of Modified Evaporation Source on The Structural characteristics of CdS Films, Vacuum, 46(12) : 1419.
- [25] de Araújo, C.B , Boudebs, G., Briois, V., Pradel, A., Messaddeq, Y., Nalin, M. 2006. Nonlinear Refractive Index Measurements in Antimony Sulfide Glass Films Using a Single Beam Nonlinear Image Technique , Optics Communications, 260 (2):723
- [26] Rajpure, K. Y. , Bhosale, C. H. 2000. Effect of Composition on the structural ,optical and electrical properties of sprayed Sb₂S₃ Thin Films Prepared From Non-Aqueous medium , Journal of Physics and Chemistry of Solids, 61(4):561
- [27] El Zawawi, I.K., Abdel-Moez, A., Terra, F. S., Mounir, M. 1998.

- Substrate Temperature Effect on the Optical and Electrical Properties of Antimony Trisulfide Thin Films ,Thin Solid Films ,324(1-2):300
- [28] Dutta, S., Chattopadhyay, S., Sarkar, A., Chakrabarti, M., Sanyal, p. and Jana, D. 2009. Role of Defects In Tailoring Structural, Electrical and Optical Properties of ZnO, Progress in Material Science, 54 : 89.
- [29] Buet, F., Olivier-Fourcade, J., Bensimon, Y., Belougne , P. 1991. Complex Impedance Study of Chalcogenide Glasses , Solid State Communications ,77(1):29

تأثير السمك على بعض الخصائص البصرية لأغشية كبريتيد الانتمون المحضرة بالترسيب بالحمام الكيميائي

نادر فاضل حبوبي* سامي سلمان جواد* صبا جميل حسن*

*الجامعة المستنصرية، كلية التربية، قسم الفيزياء

الخلاصة:

حضرت أغشية رقيقة من كبريتيد الانتمون بالترسيب بالحمام الكيميائي على قواعد زجاجية. سجل طيفي الامتصاصية والنفاذية لمدى الطول الموجي (300-900) nm. تم دراسة تأثير السمك على معامل الامتصاص، الانعكاسية، معامل الانكسار، معامل الخمود، ثابت العزل الكهربائي بجزئيه الحقيقي والخيالي. لقد وجد بان الانعكاسية، معامل الامتصاص، الجزء الحقيقي من ثابت العزل الكهربائي ومعامل الانكسار، تقل بزيادة السمك بينما ثابت العزل الخيالي يزداد.