

**Synthesis and Characterization of 8-(4-Dimethyl amino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-dione and 8-(2- hydroxy-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10- dione by reaction of 2-Oxa-spiro[3.4]octan-1,3-dione with (4-Dimethylamino-benzylidene)-(6-R-benzothiazol-2-yl)-amine and 2-[(6-R-benzothiazol-2- ylimino)-methyl]-phenol.**

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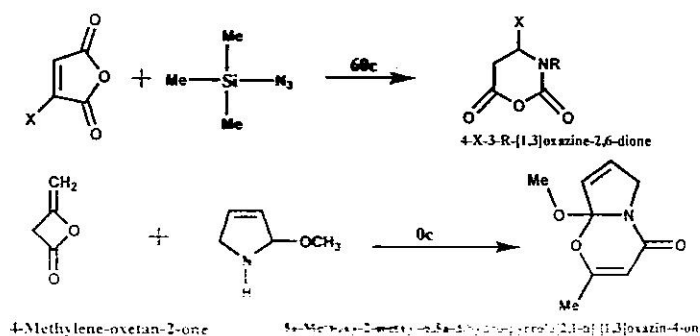
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**Abstract :-**

(4-Dimethylamino –benzylidene)-(6-R-benzothiazol-2-yl)-amine and 2-[(6-R-benzothiazol-2-ylimino)-methyl]-phenol (Schiff bases) were prepared by condensation of 6-R-2-amino benzothiazol with Salicylaldehyde and 4-dimethyl amino benzaldehyde. These Schiff bases were reacted with 2-Oxa-spiro[3.4] octane-1,3-dione by heating with reflux in dry benzene at 80°C for 3hrs. to give a six-member heterocyclic ring system; 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9- spiro[4.5] decane-6,10-diones and 8-(2-hydroxy-phenyl)-9-(6-R-benzothiazol 2-yl)-7-oxa-9- spiro[4.5]decane-6,10diones. These compounds react with pyrrolidine to give 1-(Pyrrolidine-1-carbonyl)-cyclopentanecarboxylic acid (4-dimethylamino-phenyl)-hydroxyl-methyl-(6-benzothiazol-2-yl)-amide and 1-(Pyrrolidine-1-carbonyl)-cyclopentane carboxylic acid [2-hydroxy-(2-hydroxyl-phenyl)-methyl]-(6-R-benzothiazol-2-yl)-amide. The final products were identified by their m.ps, elemental analyses, IR, and UV-Visible spectra.

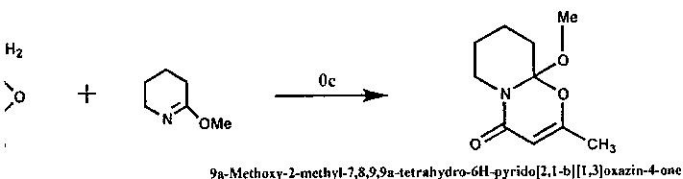
**Introduction:-**

The six –membered heterocyclic ring system: 1,3-oxazine has already been reported and thoroughly reviewed in the literature (1-4). Maleic arylmaleic and substituted maleic anhydrides react with trimethylsilyl azide to give 4- and 5-substituted "oxauraciles": dihydro-1,3-oxazine-2,6-diones (5-6)

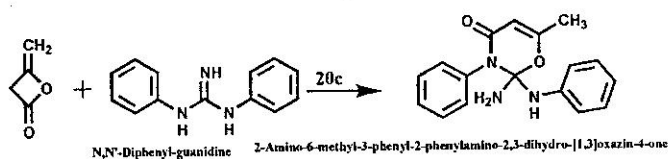


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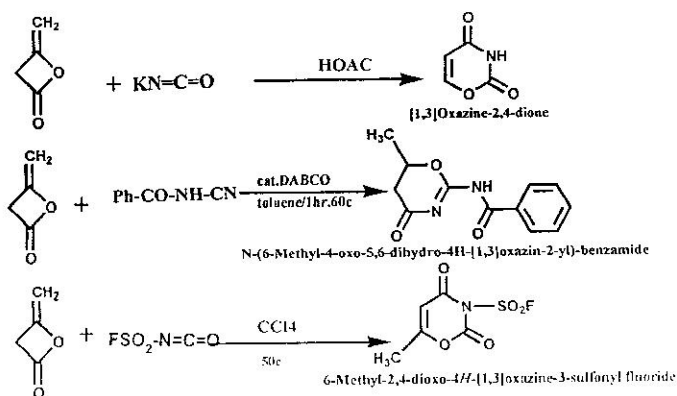
Both 2-methoxy pyrroline and 2-methoxypiperidine react with diketene under natural conditions at 0c to give the corresponding, 2-methoxy dihydro 1,3-oxazine-4-ones (7-8).



Diketene reacts with N,N-diphenyl guanidine to give the tranquilizer ketazolam and N-substituted tetrahydro-1,3-oxazine-4-one respectively (9-10).



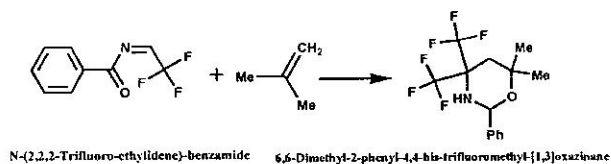
The reaction of diketene with isocyanic acid, cyanamides and fluorousulphonyl isocyanate afforded the corresponding 1,3-oxazine-2,4-diones (11-13).



Ethyl benzimideate, and ethyl butyrimidate react with diketene to give 2-ethoxy-1,3-oxazine-4-ones (14). Imines and N-acyl imines react with diketene to give tetrahydro-1,3-oxazine-4-ones (15-16).

N-acyl imines undergo [4+2] cycloaddition with both C=C- and heterodienes. For instance, isolable

bis(trifluoromethyl)acyl imine, reacts with 2,2-dimethylethylene to give 1,3-oxazine.



N-acyl immonium ions have been the most commonly used dienes to effect [4+2] cycloaddition as 4π components with substituted 1,3-butadienes. It is found that N-acylimines or immonium ions that are capable of tautomerization undergo intramolecular Diels-alder reaction to give dihydro-1,3-oxazines (17)

The reaction of N-Benzylidene 1,5-dimethyl-2-phenylpyrazolonamines (Schiff bases with Cyclopentane-1,1-dicarboxylic anhydride to give 2-aryl-3-(1,5-dimethyl-2-phenylpyrazolo)-1-(5) spirocyclopentyl tetrahydro-1,3-oxazine-4,6-diones (18).

## Experimental:-

Melting points were recorded with Gallenkamp Melting point Apparatus and were uncorrected. Elemental analysis were carried out with perkin-Elmer, 2400; CHN Elemental Analyzer. IR spectra were recorded with PYE UNICAM sp-300 Infrared spectrophotometer by using KBr disc. UV-Visible spectra were recorded (in absolute ethanol) with Shimadzu Rec-160 spectrophotometer.

(4-Dimethylamino-benzylidene)-(6-R-benzothiazol-2-yl)-amine and 2-[(6-R-benzothiazol-2-ylimino)-methyl]-phenol. (24)

A mixture of 0.01 mole of 2-amino-6-R-benzothiazol and 0.01 mole of salicylaldehyde in 20 ml absolute ethanol, was refluxed in water bath for 30 min, then left to cool in ice-water whereby yellow crystalline solid separated out. The solid was filtered, washed with 2% HCl, then with water and recrystallized twice from methanol.

Other Schiff bases were prepared following the same procedure.

**8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-spiro[4.5]decane-6,10-diones**

In a 100 ml round bottom flask equipped with a double surface condenser fitted with calcium chloride guard tube, was placed a mixture of 0.01mol of 4-Dimethylamino – benzylidene)-(6-R-benzothiazol-2-yl)-amine or 2-[(6-R-benzothiazol-2-ylimino)-methyl]-phenol and 0.01mol of 2-Oxa-spiro[3.4] octane-1,3-dione in 20ml of dry benzene . The reaction mixture was refluxed in water bath at 80c for 3hr., the solvent was then removed and the resulting solid was recrystallized from anhydrous THF.

This experimental was repeated using different (Schiff bases) in order to prepare other 8-(2-hydroxy-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-spiro[4.5]decane-6,10-diones.

**Preparation of 1-(carbonyl pyrrolidine) cyclopentane-N-(4-dimethyl amino phenyl)methylol-N-(6-R-benzothiazol-2-yl) amide.**

To a mixture of 0.005 mole of 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-spiro[4.5] decane-6,10-diones. suspended in dry THF, was added an excess(0.03 mole) of dry pyrrolidine . After 10 min of stirring the mixture at room temperature, a clear solution was obtained . The solution was refluxed at (65C°) in water bath for(45min) than left to room temprature and separated product was filtered , washed twice with (5ml) portion of dry THF and recrystallized from dioxane.

Several other derivatives of 1-(carbonyl pyrrolidine) cyclopentane-N-(4-dimethyl amino phenyl)methylol-N-(6-R-benzothiazol-2-yl) amide and 1-(carbonyl pyrrolidine) cyclopentane-N-(2-hydroxy

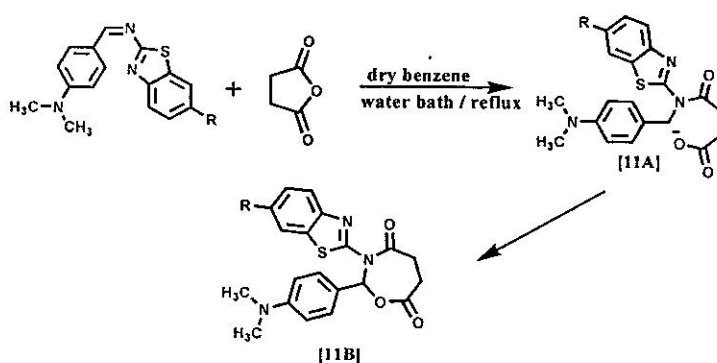
phenyl)methylol-N-(6-R-benzothiazol-2-yl) amide

were obtained following the same procedure.

**Discussion:-**

It is known that Schiff bases react smoothly with acid chlorides and anhydrides to give the corresponding addition products<sup>(19-21)</sup> .

In this paper, the reaction of 2-Oxa-spiro[3.4] octane-1,3-dione with (4-Dimethylamino –benzylidene)-(6-R-benzothiazol-2-yl)-amine to give the dipolar intermediate [11A] which collapses to the 6- membered hetrocyclic ring system.[11B] is presented.



**[Schem 1]**

This is indicated by the appearance of the characteristic C=O (lacton-lactam) absorption band at 1700cm<sup>-1</sup> in the IR spectra of addition products[11B].

It is impressive to note that the two absorption band at (1800-1950)cm<sup>-1</sup> in the IR spectra of pure 2-Oxa-spiro[3.4] octane-1,3-dione (anhydride) have disappeared when the anhydride became part of the 6-membered ring system of the 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones and 8-(2-hydroxy-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones .The new absorption bands of the (C=O) group in the IR spectra of the addition products [11B] appear at (1670-1700)cm<sup>-1</sup>. this attributed to the fact that the structures of

the addition products are combination of the lacton-lactam structure.

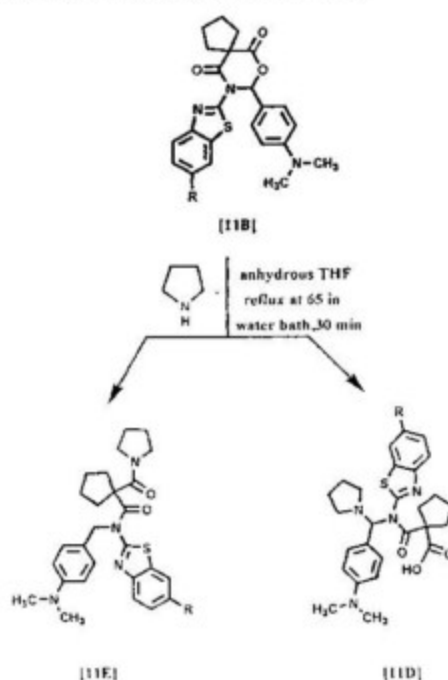
The UV spectra of 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones and 8-(2-hydroxyphenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones show absorption maxima at (240-310)nm, and at (310-445)nm due to charge transfer of the aryl group and the cyclic 6-membered structure [11B].

8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones and 8-(2-hydroxyphenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones are identified by their m.p.s, elemental analysis (table 1), IR spectra (table 2) and UV spectra (table 3). It is noticeable that the values of C-H str. (benzylic) absorption bands are rather high. This is in fact explained by the shift toward longer wavelength, that takes place when the benzylic carbon is linked to three electron-withdrawing groups, phenyl, O and N in the titled compounds.

The reaction of 2-Oxa-spiro[3.4]octane-1,3-dione with various Schiff bases is a sort of cycloaddition reaction. Cycloaddition includes ring formation which results from the addition of bonds to either  $\delta$  or  $\pi$  with formation of new  $\delta$  bonds. This class of reactions and its reverse encompasses a large number of individual types. Huisgen<sup>(22)</sup> has formulated a useful classification of diverse cycloaddition in terms the number of the new  $\delta$  bond, the ring size of the product, and the number of atoms in the components taking part in the cycloaddition. This cycloaddition reaction is classified as a 2 + 5-7, and it is the first cycloaddition of this type, although in principle, one would predict that the butadienyl cation might add to an olefin

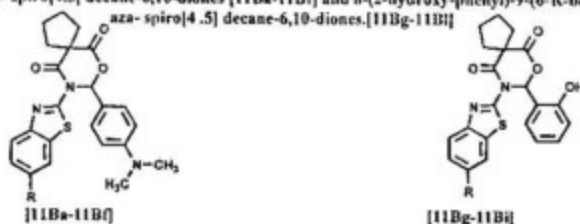
through a  $(4n+2)$  transition state to yield the cyclohexenyl cation<sup>(23)</sup>.

In an attempt to cleave the 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones and 8-(2-hydroxyphenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-diones are treated with anhydrous pyrrolidene to give the corresponding open-chain 1-(Pyrrolidine-1-carbonyl)-cyclopentanecarboxylic acid (4-dimethylamino-benzyl)- (6-benzothiazol-2-yl)-amide and 1-(Pyrrolidine-1-carbonyl)-cyclopentanecarboxylic acid (2-hydroxybenzyl)- (6-R-benzothiazol-2-yl)-amide. Apparently, the reaction involves an acyl-oxygen cleavage of the  $\gamma$ -lactone ring, while the anilide N-C=O linkage unaffected under these conditions. Since none of the two nitrogen atoms in the resulting products, carries hydrogen, reclosure to the cyclic diamides is not expected. It is also found that the products are insoluble in water, dil. HCl and dil. NaOH solution, which supports structure [11D], rather than the one formed by an alkyl-oxygen cleavage [11E], an amino acid which is expected to be soluble in acidic and basic medium.



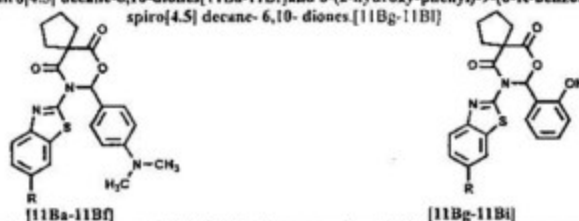
Scheme 2

Table (1) : Some physical and experimental properties of 8-(4-Dimethylamino-phenyl)-9-(6- R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones [11Ba-11Bf] and 8-(2-hydroxy-phenyl)-9-(6- R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones.[11Bg-11Bi]



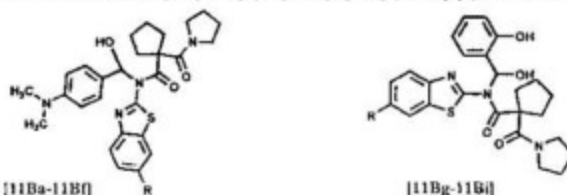
Comp.	R	M.P/ C	Yield%	Molecular formula	Calc.			Found		
					C	H	N	C	H	N
11Ba	CH <sub>3</sub>	215	77	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub> S	66.18	5.79	9.65	66.20	7.88	9.54
11Bb	4,6 -CH <sub>3</sub>	243	69	C <sub>22</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub> S	66.79	6.05	9.35	66.90	5.96	9.33
11Bc	NO <sub>2</sub>	188	60	C <sub>23</sub> H <sub>12</sub> N <sub>4</sub> O <sub>2</sub> S	59.22	4.75	12.01	59.34	4.80	11.99
11Bd	OCH <sub>3</sub>	209	54	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> O <sub>4</sub> S	63.84	5.58	9.31	63.95	5.51	9.26
11Be	Cl	230	61	C <sub>23</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> SCl	60.59	4.86	9.22	60.60	4.89	9.21
11Bf	Br	255	65	C <sub>23</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> SBr	55.20	4.43	8.40	55.35	4.31	8.45
11Bg	CH <sub>3</sub>	237	72	C <sub>27</sub> H <sub>26</sub> N <sub>2</sub> O <sub>4</sub> S	64.69	4.94	6.86	64.81	5.02	6.77
11Bh	4,6 -CH <sub>3</sub>	222	66	C <sub>25</sub> H <sub>22</sub> N <sub>2</sub> O <sub>4</sub> S	65.38	5.25	6.63	65.45	5.15	6.57
11Bi	NO <sub>2</sub>	211	61	C <sub>23</sub> H <sub>12</sub> N <sub>4</sub> O <sub>2</sub> S	57.40	3.90	9.56	57.53	4.05	9.61
11Bj	OCH <sub>3</sub>	260	55	C <sub>27</sub> H <sub>26</sub> N <sub>2</sub> O <sub>6</sub> S	62.25	4.75	6.60	62.33	4.66	6.53
11Bk	Cl	256	69	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>4</sub> SCl	58.81	4.00	6.53	58.76	4.10	6.56
11Bl	Br	245	63	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>4</sub> SBr	53.29	3.62	5.92	53.45	3.76	5.82

Table (2): The major IR absorptions bands (cm<sup>-1</sup>) of 8-(4-Dimethylamino-phenyl)-9-(6- R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones[11Ba-11Bf]and 8-(2-hydroxy-phenyl)-9-(6- R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10- diones [11Bg-11Bi]



Comp.	O-H str.	C-H str.	C=O str Lacton	C-Ostr. Lactam	C-Cstr. Aromatic	C-N str.	C-O bend Lacton	Others
11Ba	-----	3220	1670	1650	1600,1570,1470	1360,1020	1290	
11Bb	-----	3230	1670	1655	1580,1550,1460	1350,1010	1280	
11Be	-----	3220	1680	1640	1590,1560,1470	1380,1030	1300	1365,1540 NO <sub>2</sub>
11Bd	-----	3225	1690	1640	1600,1560,1460	1375,1010	1310	1220 C-O-C ether
11Be	-----	3220	1690	1645	1580,1450,1460	1360,1030	1290	790 C-Cl
11Bf	-----	3230	1680	1650	1570,1545,1475	1375,1025	1285	630 C-Br
11Bg	3430	3230	1670	1640	1600,1560,1450	1365,1020	1275	
11Bh	3440	3235	1660	1650	1600,1580,1470	1380,1035	1280	
11Bi	3440	3235	1680	1635	1600,1560,1480	1365,1035	1270	1350,1550 NO <sub>2</sub>
11Bj	3430	3235	1690	1635	1590,1550,1480	1370,1020	1290	1210 C-O-C ether
11Bk	3435	3250	1680	1645	1580,1530,1465	1375,1040	1300	770 C-Cl
11Bl	3430	3225	1670	1650	1590,1560,1470	1370,1040	1300	590 C-Br

Table (3) : Some physical and experimental properties of 1-(Pyrrolidine-1-carbonyl)- cyclopentancarboxylicacid(4-dimethylamino- phenyl)- hydroxyl-methyl-(6-benzothiazol-2-yl)-amide and 1-(Pyrrolidine-1-carbonyl)-cyclopentane carboxylic acid(2- hydroxy-2-hydroxyl-phenyl)-methyl)-(6- R- benzothiazol-2-yl)-amide.



Comp.	R	M.P/C	Yield%	Molecular formula	Calc.			Found		
					C	H	N	C	H	N
11Ea	CH <sub>3</sub>	237	66	C <sub>23</sub> H <sub>22</sub> N <sub>2</sub> O <sub>5</sub> S	66.64	7.33	10.72	66.70	7.32	10.82
11Eb	4,6 -Cl <sub>2</sub>	264	69	C <sub>24</sub> H <sub>16</sub> N <sub>2</sub> O <sub>5</sub> S	66.89	6.97	10.76	67.00	6.88	10.61
11Ec	NO <sub>2</sub>	220	72	C <sub>23</sub> H <sub>12</sub> N <sub>4</sub> O <sub>5</sub> S	60.32	5.81	13.03	60.44	5.92	10.11
11Ed	OCH <sub>3</sub>	263	58	C <sub>23</sub> H <sub>22</sub> N <sub>2</sub> O <sub>6</sub> S	64.34	6.56	10.72	64.50	6.55	10.67
11Ee	Cl	197	62	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> SCl	61.53	5.93	10.63	61.60	6.02	10.55
11Ef	Br	189	60	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> SBr	56.74	5.47	9.80	56.83	5.53	9.76
11Eg	CH <sub>3</sub>	231	67	C <sub>26</sub> H <sub>26</sub> N <sub>2</sub> O <sub>5</sub> S	65.11	6.09	8.76	65.23	6.00	8.85
11Eh	4,6 -CH <sub>3</sub>	244	64	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> O <sub>5</sub> S	65.70	6.33	8.51	65.73	6.38	8.58
11Ei	NO <sub>2</sub>	237	67	C <sub>23</sub> H <sub>12</sub> N <sub>4</sub> O <sub>5</sub> S	58.81	5.13	10.97	58.93	5.07	11.04
11Ej	OCH <sub>3</sub>	250	55	C <sub>26</sub> H <sub>22</sub> N <sub>2</sub> O <sub>6</sub> S	63.01	5.90	8.48	63.00	6.03	8.52
11Ek	Cl	246	71	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> SCl	60.05	5.24	8.40	60.11	5.30	8.56
11El	Br	227	73	C <sub>23</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> SBr	55.15	4.81	7.72	55.16	4.80	7.84

Table(4): The major IR absorptions bands ( $\text{cm}^{-1}$ ) of 1-(Pyrrolidine-1-carbonyl)-cyclopentanecarboxylic acid(4-dimethylamino-phenyl)-hydroxyl-methyl-(6-benzothiazol-2-yl)-amide[11Ea-11Ej] and 1-(Pyrrolidine-1-carbonyl)-cyclopentane carboxylic acid [2-hydroxy-(2-hydroxyl-phenyl)-methyl]-(6-R-benzothiazol-2-yl)-amide.[11Eg-11Ei]



Comp.	O-H str. Olefine	O-H str. phenol	C=O str. $\Delta$ Amide	C=Cstr. Aromatic	C-H bend Benzylic	C-Ostr. Alcohol	C-S str.	Others
11Ea	3450	-----	1680	1580,1490	1400	1360	1240	
11Eb	3455	-----	1670	1590,1520	1410	1350	1250	
11Ec	3460	-----	1670	1580,1510	1415	1365	1245	1355,1540 $\text{NO}_2$
11Ed	3460	-----	1660	1590,1480	1385	1350	1260	1220 C-O-C ether
11Ee	3480	-----	1680	1585,1530	1390	1345	1235	720 C-Cl
11Ef	3440	-----	1650	1590,1520	1400	1335	1240	630 C-Br
11Eg	3490	3355	1665	1580,1540	1410	1330	1250	
11Eh	3450	3345	1685	1570,1510	1420	136	1245	
11Ei	3470	3390	1675	1580,1480	1400	1355	1260	1345,1545 $\text{NO}_2$
11Ej	3460	3340	1670	1570,1530	1395	1350	1255	1210 C-O-C ether
11Ek	3450	3345	1685	1590,1510	1380	1345	1230	710 C-Cl
11El	3480	3360	1670	1590,1485	1390	1330	1250	590 C-Br

Table ( 5 ) : The UV-Visibl absorption maxima  $\lambda/\text{nm}$  of 8-(4-Dimethylamino-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones ,8-(2-hydroxy-phenyl)-9-(6-R-benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5] decane-6,10-diones. [11Ba-11Bi] and 1-(Pyrrolidine-1-carbonyl)-cyclopentanecarboxylic acid(4-dimethylamino-phenyl)-hydroxyl-methyl-(6-benzothiazol-2-yl)-amide and 1-(Pyrrolidine-1-carbonyl)-cyclopentane carboxylic acid [2-hydroxy-(2-hydroxyl-phenyl)-methyl]-(6-R-benzothiazol-2-yl)-amide.[11Ea-11Ej]

compound	UV-Visibl absorption maxima $\lambda/\text{nm}$ of oxazepane [11Ba-11Bi]	Comp.	UV-Visibl absorption maxima $\lambda/\text{nm}$ of anilid - pyrrolidides (Amide) [11Ea-11Ej]
11Ba	350,270,230,220	11Ea	385,310,270,254,222
11Bb	325,265,245,225	11Eb	359,310,268,248,229
11Be	345,290,260,230,224	11Ec	370,300,268,245,225
11Bd	375,268,259,234,222	11Ed	355,320,268,240,225
11Be	340,280,240,228	11Ee	355,277,252,244,226
11Bf	365,340,275,247,226	11Ef	349,305,267,247,223
11Bg	360,285,250,227	11Eg	359,300,280,254,224
11Bh	340,330,276,240,220	11Eh	344,295,266,242,228
11Bi	338,278,247,227	11Ei	370,276,245,236,228
11Bj	363,255,241,233,221	11Ej	350,289,266,245,227
11Bk	335,278,253,224	11Ek	351,273,265,245,221
11Bl	359,269,249,238,225	11El	359,309,271,248,220

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تحضير ودراسة الخواص التركيبية ل- ٨-(٤-ثنائي أمينو-فنييل)-٩-(٦-معوض-بنزو ثيازول-٢-يل)-٧-أوكسا-٩-أزا-سبيرو [٤,٥] ديكان-٦,١٠-دايون و ٨-(٢-هيدروكسي-فنييل) -٩-(٦-معوض-بنزو ثيازول-٢-يل)-٧-أوكسا-٩-أزا-سبيرو [٤,٥] ديكان-٦,١٠-دايون من تفاعل ٢-أوكسا-سبيرو [٤,٥] اوكتان-١,٣-دايون مع (٤-ثنائي مثيل أمينو-بنزيل دين)-٦-(معوض-بنزو ثيازول-٢-يل)-أمين و ٢ [(٦-معوض-بنزو ثيازول-٢-يل أمينو)-مثيل]-فينول.

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الخلاصة :-

تم تحضير عدد من (قواعد شيف) بتكاتف ٦-معوض-٢-أمينو بنزو ثيازول مع الساليسيل الديهايد و ٤-ثنائي مثيل أمينو بنزو ثيازول. فوعلت قواعد شيف هذه مع ٢-أوكسا-سبايرو [٤,٣] اوكتان-١ و ٣-دايون بالتصعيد في البنزين الجاف عند درجة ٨٠م لثلاث ساعات لتعطي نظام حلقي سداسي غير متجانس: ٨-(٤-ثنائي مثيل أمينو-فنييل)-٩-(٦-معوض-بنزو ثيازول-٢-يل)-٧-أوكسا-٩-سبايرو [٥,٤] ديكان-٦ و ١٠-دايون و ٨-(٢-هيدروكسي-فنييل)-٩-(٦-معوض-بنزو ثيازول-٢-يل)-٧-أوكسا-٩-سبايرو [٥,٤] ديكان-٦ و ١٠-دايون. تم مفاعلة الأخيرة مع البيرو لدين فأعطت مشتقات ١-(بيرو لدين-١-كاربونيل)-سايكلوبنتان كاربو كسيلك أسد(٤-ثنائي مثيل-بنزيل)-٦-(معوض-بنزو ثيازول-٢-يل)-اميد و ١-(بيرو لدين-١-كاربونيل)-سايكلوبنتان كاربو كسيلك أسد(٢-هيدروكسي-بنزيل)-٦-(معوض-بنزو ثيازول-٢-يل)-اميد. وقد شخصت هذه المركبات بواسطة درجات انصهارها وتحليل العناصر الدقيق وأطياف UV-IR.